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Problem Solving continued Boyle 's law can be expressed by the followmg mathematical
equation. = Pressure under Volume under Pressure under Volume under thefirst set of X

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the first set of = the second set X the second set conditions conditions of conditions
of conditions General Plan for Solving Boyle's Law Problems Rearrange the equation $P_1 V_1 = P_2 V_2$

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Skills Worksheet Problem Solving

Holt ChemFile: Problem-Solving Workbook 3 Conversions Name Class Date Problem Solving continued The correct conversion factor is the one that when multiplied by the given quantity causes the units to cancel. COMPUTE 45.5 g calcium chloride 100 1 0 g mg 45 500 mg calcium chloride EVALUATE Are the units correct? Yes; milligrams are the desired units.

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Solving continued Sample Problem 1 Ammonia is made industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$.

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Holt ChemFile: Problem-Solving Workbook 192 Stoichiometry of Gases Name Class Date

Problem Solving continued Practice 1. Complete the table below using the following equation,

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which represents a reaction that produces aluminum chloride. $2\text{Al}(s) + 3\text{Cl}_2(g) \rightarrow 2\text{AlCl}_3(s)$
Mass Al Volume Cl_2 Conditions Mass AlCl_3 a. excess ?

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Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is
about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain ...

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