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pK<sub>BH</sub> + (1) = 2,38 (25°C, ) pK<sub>a</sub> (1) = 9,39 (25°C, ) . (II) . 1,4- -2- .

I-  
Empirical formula (Hill's system for organic substances): C 1 8 H 3 6 O 2 Structural formula as text: CH<sub>3</sub>(CH<sub>2</sub>)<sub>16</sub>COOH ... Dictionary of organic compounds. - Vol. 3, naphthacarbazole - zygadenine. - ?.: ???, 1949. - pp. 637;

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Empirical formula (Hill's system for organic substances): C<sub>16</sub>H<sub>36</sub>IN Structural formula as text: (C<sub>4</sub>H<sub>9</sub>)<sub>4</sub>NI Molar/atomic mass: 369.368 Melting point (°C): 147 Solubility (g/100 g of solvent): 1-butanol: 62.28 (25°C) water: soluble methanol: 256.5 (25°C) chloroform: sparingly soluble ethanol: soluble . References:

## **tetrabutylammonium iodide**

Empirical formula (Hill's system for organic substances): C<sub>18</sub>H<sub>35</sub>AgO<sub>2</sub> Structural formula as text: CH<sub>3</sub>(CH<sub>2</sub>)<sub>16</sub>COOAg ... Dictionary of organic compounds. - Vol. 3, naphthacarbazole - zygadenine. - ?.: ???, 1949. - pp. 637;

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Empirical formula (Hill's system for organic substances): C<sub>18</sub>H<sub>36</sub>O<sub>7</sub> Molar/atomic mass: 344.33 Melting point (°C): 196 Solubility (g/100 g of solvent): water: insoluble diethyl ether: soluble chloroform: very soluble ethanol: soluble . Dissociation: pK<sub>a</sub> (1) = 4.4 (25°C, water) pK<sub>a</sub> (2) = 8.8 (25°C, water)

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