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MME

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dye becomes less effective as the hydrogen peroxide decomposes forming water and oxygen.

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kJ mol⁻¹ B E A = 50 kJ mol⁻¹
and $\Delta H = +100$ kJ mol⁻¹ C E A
= 100 kJ mol⁻¹ and $\Delta H = +50$
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mol⁻¹ and $\Delta H = -50$ kJ mol⁻¹

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CfE Higher Chemistry Unit 1:
Chemical Changes and
Structure

In general: Average rate =
change in variable change in
time It is worth mentioning
at this stage that the unit
of concentration of a

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solution is measured as moles per litre. In previous courses this may have been shortened to mol/ or even as mol/dm³At this level the same unit will always be written as mol⁻¹.

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o $a = 1, b = 3, c = 2$ and d

$= 4$ o $a = 1, b = 3, c = 2$

and $d = 2$ o $a = 1, b = 2, c$

$= 2$ and $d = 2$. National 5

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Unit 1 - Chemical Changes &

Structures. 6 - The Mole. a)

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Gram Formula Mass (GFM) The gram formula mass (GFM) of a substance is known as the mass of one mole.

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