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Adding the number of protons (12) and the number of neutrons (11) together, reveals the mass of the atom to be 23. Many excited state configurations are possible. A few examples include: 2-7-4, 2-5-6, 1-8-4. They contain different numbers of neutrons. $(19.99 \times .909) + (20.99 \times 0.03) + (21.99 \times 0.88)$ 20 $(12.00 \times 9.93) + (13.00 \times 1.07)$ 100

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ANSWERS To TOPIC 12 Pret is an -c-o-c-c- c- CH O-C-CH CH CH 41. Butanol or n-butanol 42. [1] Allow 1 credit. Examples of a 1-credit response: 12. 3 43. 36 Review Questions 10. 13. 15. 16. 19. 22. 25. 28. 31. 34. 40. 43. 46. 49. 52. 55. 1 2 11. 3 14. 3 -c- c- c- H c c- 18. 21. 24. 30 33. 36. 39. 42. 45. 48. 51. 54. 57. (a) _ Ole + 21gBi (b) _qe + lggBa 222

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Topic 12 - Acids, Bases and Buffers - A-Level Chemistry

A solution was made by dissolving 12.5 g of Na₃PO₄ in 125.00 mL of water. The volume of the resulting solution was 128 mL. Calculate the molarity and molality of the solution.

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(12.00 x 9.93) + (13.00 x 1.07) 100 Both isotopes contain 6 protons but different numbers of neutrons. Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 protons and 7 neutrons. CHEMISTRY ANSWER KEY 27 Na Mg Al 1s22s22p63s23p1 Si 1s22s22p63s23p2 P S 1s22s22p63s23p4

Quia

12.02 They contain different numbers of neutrons. (19.99 x .909) + (20.99 x 0.03) + (21.99 x 0.88) 20 (12.00 x 9.93) + (13.00 x 1.07) 100 Both isotopes contain 6 protons but different numbers of neutrons. Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 Na Mg 1s22s22p63s2

St. Francis Preparatory School

12 + H₂O → H₂O + + 1- The vertical columns of the table should have elements with similar chemical properties. If the elements were arranged by atomic mass, several pairs of elements, such as Ar and K, would be out of position in regards to properties. Both groups follow the octet rule. Group I members achieve an octet by losing an electron,

Answers to Topic 5 Periodic Table from Review Book

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Chem Key - Chemistry The Physical Setting Answer Key ...

Review Topics and Practice Evaluation Questions These are the topics a student should know fairly well in order to skip Chemistry 122 and enter directly into Chemistry 123 or 128. The more difficult topics are marked with an asterisk.

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