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of the oxygen (N, O, F, and Cl have high values) • bond angle of water = 105°.

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Cotton 2. Section 15.1 Water and its Properties OBJECTIVES: –Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

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hydrates are not very strong, so the water is easily lost and regained.  
Work Step by Step.

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### Systems. This creates: Surface

Tension. The \_\_\_\_\_ force, or pull, that tends to minimize the surface area of the liquid

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Chemistry, Chapter 15, Water and Aqueous Systems. surface tension. surfactant. aqueous solution. solvent. the inward force or pull that tends to minimize the surface ar.... any substance that interferes with hydrogen bonding between wa.... is water that contains dissolved

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substances. the dissolving medium in  
a solution.

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445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. Water in the Liquid State (pages 445–447)

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PROPERTIES (pages 445–449)  
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source:eur-lex.europa.eu He may  
want to stretch himself once a worker

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knows his efforts do not go unnoticed. For instance, if he knows his performance will be judged based on achievement of a target, he will work harder to achieve it.

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compounds are electrolytes because they dissociate into ions and conduct an electrical current in aqueous solution or in the molten state.

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Practice Problems: Which  
tensions of most other liquids? Which  
type of mixture(s) exhibit the ...

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