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~~pH, pOH, H<sub>3</sub>O<sup>+</sup>, OH<sup>-</sup>, K<sub>w</sub>, K<sub>a</sub>, K<sub>b</sub>, pK<sub>a</sub>, and pK<sub>b</sub> Basic Calculations Acids and Bases Chemistry Problems 12 4 Practice Problems Chemistry~~  
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## Solved: PRACTICE PROBLEMS 12.4 Reactions Of Alcohols, Thio ...

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Practice Problems 1. Identify each of the following as a formula of an organic or inorganic compound: a. KCl b. C<sub>4</sub>H<sub>10</sub> c. C<sub>2</sub>H<sub>6</sub>O d. H<sub>2</sub>SO<sub>4</sub> e. CaCl<sub>2</sub> f. C<sub>3</sub>H<sub>7</sub>Cl 2. Identify each of the following as a

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formula of an organic or inorganic compound: a.  $C_6H_{12}O_6$  b.  $K_3PO_4$  c.  $I_2$  d.  $C_2H_6S$  e.  $C_{10}H_{22}$  f.  $C_4H_9Br$  3.

### Chapter 12 Introduction to Organic Chemistry: Hydrocarbons

Figure 12.1.4 Dissolution of 1 mol of an Ionic Compound In this case, dissolving 1 mol of  $(NH_4)_2Cr_2O_7$  produces a solution that contains 1 mol of  $Cr_2O_7^{2-}$  ions and 2 mol of  $NH_4^+$  ions. (Water molecules are omitted from a molecular view of the solution for clarity.)

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12-4 Practice Problems. Show work. 1. When a 12.8-g sample of  $KCl$  dissolves in 75.0-g of water in a calorimeter, the temperature drops from  $31.0^\circ C$  to  $21.6^\circ C$ . Calculate  $\Delta H$  for the process.  $KCl$   $K^+$   $Cl^-$ . 6. When a 19.2-g sample of  $KCN$  dissolves in 65.0-g of water in a calorimeter, the temperature drops from  $28.1^\circ C$  to  $15.4^\circ C$ .

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Chemistry 12 Worksheet 4-4-Ka & Kb Calculations Chemistry 12-Worksheet 4-4-Ka & Kb Calculations Page 2 of 4 Pages 3. The pH in a 0.25 M solution of the acid  $HBrO$  is 4.65. Using this, calculate the value of  $K_a$  for the acid  $HBrO$ . (4 marks) Answer \_\_\_\_\_ 4. The pH in a solution of benzoic acid is 2.355. Determine the molar concentration of the ...

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*Chapter 17 - Questions and Problems - Page 824: 17.4*

Kresen. Chemistry Chapter 12 Recommended Problems. Covalent bonding occurs in both molecul... Silicon is the fundamental component of... Silicon is the fundamental component of... Which kind of attractive forces exist b... The molecules in molecular solids are held together by weak in... covalent-network solid.

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